

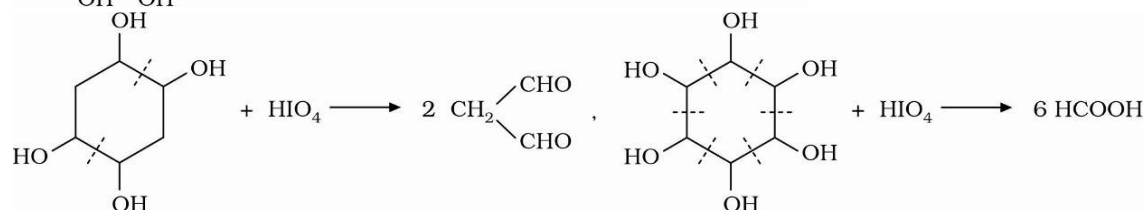
Daily Tutorial Sheet-11

Numerical Value Type for JEE
Main

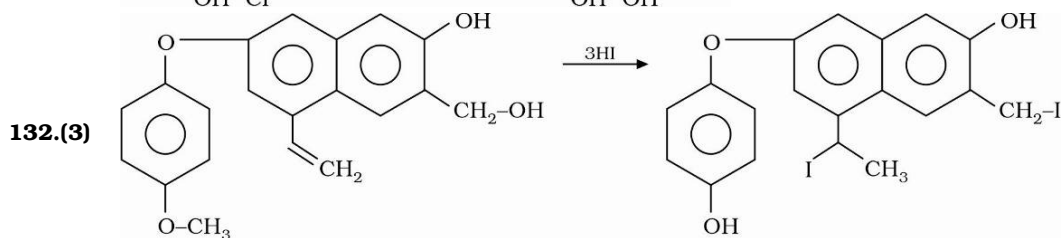
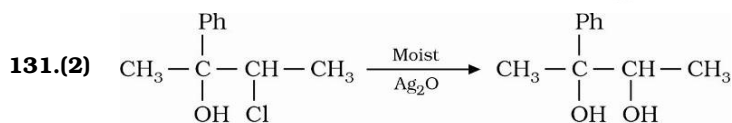
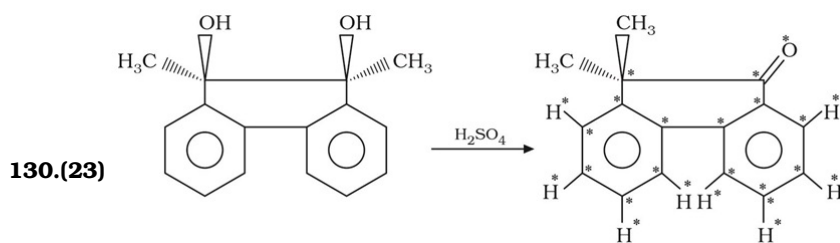
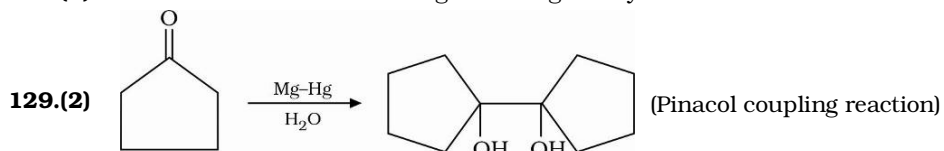
126.(3) $\text{CH}_3 - \overset{\text{O}}{\parallel} \text{C} - \overset{\text{O}}{\parallel} \text{C} - \text{OH}$ will not give reaction with HIO_4

127.(12) $\text{CH}_3 - \underset{\text{OH}}{\underset{|}{\text{CH}}} - \overset{\text{O}}{\parallel} \text{C} - \text{CH}_3 + \text{HIO}_4 \longrightarrow \text{CH}_3 - \text{CHO} + \text{CH}_3\text{COOH}$

$\text{CH}_3 - \underset{\text{OH}}{\underset{|}{\text{CH}}} - \underset{\text{OH}}{\underset{|}{\text{CH}}} - \text{CH}_2 - \text{CH}_3 + \text{HIO}_4 \longrightarrow \text{CH}_3 - \text{CHO} + \text{CH}_3 - \text{CH}_2 - \text{CHO}$

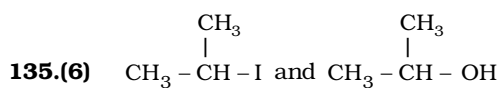


128.(1) Pinacol - Pinacolone rearrangement is given by vicinal diol



133.(1) Tertiary alcohol will be more reactive than secondary alcohol for Lucas reagent.

134.(3) Primary alcohol, Iodide and nitro compound will give red colouration in Victor-Mayer test.

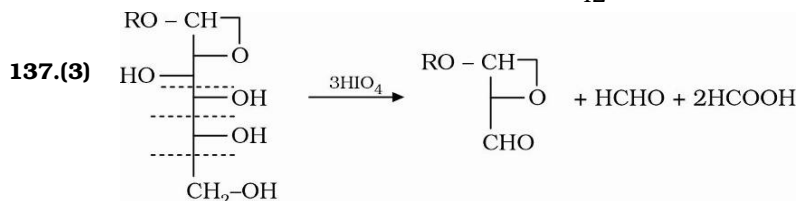


Can not be distinguished because they both are secondary groups.

136.(5) $\text{CH}_3 - \overset{\text{O}}{\parallel} \text{C} -$ group having mass = 43.

→ One acetyl group is added by removal of 1 hydrogen atom

→ So, number of hydroxyl group are $\frac{390 - 180}{42} = 5$.



138.(3) $\text{C}_3\text{H}_8\text{O}_3 + \text{CH}_3\text{MgI} \longrightarrow \text{CH}_4$

0.092	67
92	22400
= 0.001 mole	= 0.003 mole

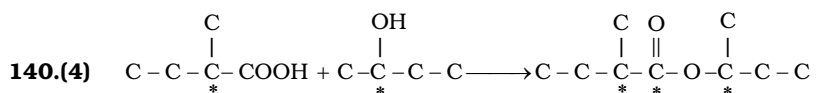
Therefore, 3 active hydrogen atoms are present.

139.(5) 1. $\text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{OH}$

2. $\begin{array}{c} \text{C} - \text{C} - \text{C} - \text{C} - \text{OH} \\ | \\ \text{C} \end{array}$

3. $\begin{array}{c} \text{HO} - \text{C} - \overset{*}{\text{C}} - \text{C} - \text{C} \\ | \\ \text{C} \end{array} \quad (2 \text{ stereoisomer})$

4. $\begin{array}{c} \text{C} \\ | \\ \text{C} - \text{C} - \text{C} - \text{OH} \\ | \\ \text{C} \end{array}$



→ (R,R)

→ (R,S)

→ (S,R)

→ (S,S)

Because of two chiral center in the final molecule, four optically active compounds are present.